

## First-principles calculations of thermodynamic properties of superhard orthorhombic $\beta$ -BC<sub>2</sub>N

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**Abstract.** We investigate the thermodynamic properties of the potential superhard orthorhombic structure boron-carbonitride  $\beta$ -BC<sub>2</sub>N by using *ab initio* plane-wave pseudopotential density functional theory method within both local density approximation (LDA) and generalized gradient approximation (GGA). The lattice parameters ( $a$ ,  $b$  and  $c$ ), equilibrium volume  $V$ , bulk modulus  $B_0$  and its pressure derivative  $B_0'$  have been calculated, and compared with those of diamond and cubic boron nitride (c-BN). The obtained results are in excellent agreement with the available experimental data and other theoretical results. Through the quasi-harmonic Debye model, we also investigate the thermodynamic properties of  $\beta$ -BC<sub>2</sub>N. The variation of the thermal expansion  $\alpha$ , the heat capacity  $C_V$  and the Grüneisen parameter  $\gamma$  with pressure  $P$  and temperature  $T$ , as well as the pressure-normalized volume ( $P$ - $V_n$ ) and the pressure-bulk modulus ( $P$ - $B$ ) relationship of  $\beta$ -BC<sub>2</sub>N are obtained systematically.

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**Key words:** density functional theory, thermodynamic properties, BC<sub>2</sub>N

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## 1 Introduction

Superhard materials are of important in modern high-pressure science and technology due to their outstanding properties, such as ultra-incompressible, high elastic modulus and hardness, scratch resistance, and so on. It is well-known that diamond and cubic boron nitride (c-BN) are considered to rank first and second among the known superhard materials, respectively. By mixing diamond with c-BN, one may create a new pseudo-diamond BC<sub>2</sub>N alloy compound, which is harder than c-BN. Such a kind of alloys are expected to be thermally and chemically more stable than diamond and harder than c-BN.

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The ternary boron-carbonitride systems, which are isoelectronic with diamond and c-BN, have been attracted more attention because the theoretical prediction that  $\beta$ - $C_3N_4$  compound should have a hardness comparable to diamond [1]. However, most of the synthesized B-C-N compounds have the turbostratic or amorphous structures so far. A majority of the previous experimental studies on this topic originated from the isoelectronic carbon and c-BN viewed should have the similar crystal structures to diamond and c-BN. The theoretically studied structures are all focus on high dense phases, such as cubic  $BC_2N$  [2–9], hexagonal  $BC_2N$  [10–14], and chalcopyrite  $BC_2N$  [15].

The ternary B-C-N compound,  $BC_2N$ , as a new ultra-hard material also known as diamond and c-BN, has an orthorhombic crystal structure belongs to space group  $Pmm2$ . Therefore, it likely supersede the expensive diamond in various potential applications. Recently, the lattice dynamics and thermodynamics of orthorhombic  $\beta$ - $BC_2N$  have been investigated by Cheng *et al.* [16]. To our knowledge, there are no report on the thermodynamic properties of orthorhombic  $\beta$ - $BC_2N$  under high pressure and temperature.

In this work, we investigate the structure and thermodynamic properties of  $\beta$ - $BC_2N$  through the Cambridge Serial Total Energy Package (CASTEP) program [17, 18] and the quasi-harmonic Debye model [19]. The results obtained are well consistent with the available experimental data and other theoretical results. The paper is organized as follows: In Section 2, we make a brief review of the theoretical method. The results and some discussion are presented in Section 3. Finally, the conclusions derived from our calculations are summarized in Section 4.

## 2 Theoretical method

### 2.1 Total energy electronic structure calculations

In our electronic structure calculations, we adopt the non-local ultrasoft pseudopotential (USPP) introduced by Vanderbilt [20] for the interactions of the electrons with the ion cores. The exchange and correlation terms are described by both the local density approximation (LDA-CAPZ) proposed by Vosko *et al.* [21] and the generalized gradient approximation (GGA-PBE) proposed by Perdew *et al.* [22]. The electronic wave functions are expanded in a plane wave basis set with energy cut-off of 550 eV. Pseudo-atom calculations are performed for B ( $2s^22p^1$ ), C ( $2s^22p^2$ ) and N ( $2s^22p^3$ ). For the Brillouin-zone  $k$ -point sampling, we use the Monkhorst-Pack mesh with  $10 \times 10 \times 7$   $k$ -points for both LDA and GGA calculations, where the self-consistent convergence of the total energy is  $5.0 \times 10^{-7}$  eV/atom. These parameters are sufficient in leading to well converged total energy and geometrical configurations. All total energy electronic structure calculations are implemented through the CASTEP code [17, 18].