## The Traveling Wave of Auto-Catalytic Systems-Monotone and Multi-Peak Solutions

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**Abstract.** This article studies propagating wave fronts of a reaction-diffusion system modeling an isothermal chemical reaction  $A+2B \rightarrow 3B$  involving two chemical species, a reactant A and an auto-catalyst B, whose diffusion coefficients,  $D_A$  and  $D_B$ , are unequal due to different molecular weights and/or sizes. Explicit bounds  $c_*$  and  $c^*$  that depend on  $D_B/D_A$  are derived such that there is a unique travelling wave of every speed  $c \ge c^*$  and there does not exist any travelling wave of speed  $c < c_*$ . Furthermore, the reaction-diffusion system of the Gray-Scott model of  $A+2B \rightarrow 3B$ , and a linear decay  $B \rightarrow C$ , where C is an inert product is also studied. The existence of multiple traveling waves which have distinctive number of local maxima or peaks is shown. It shows a new and very distinctive feature of Gray-Scott type of models in generating rich and structurally different traveling pulses.

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Key words: Qubic autocatalysis, travelling wave, minimum speed, Gray-Scott, multi-peak waves.

## 1 Introduction

Autocatalytic chemical reaction of the form

 $A + nB \rightarrow (n+1)B$  with rate  $kab^n$  and n = 1,2

between two chemical species *A* and *B*, appears in many chemical wave models of excitable media from the idealized Brusselator to real-world clock reactions such as Belousov-Zhabotinsky reaction, the Briggs-Rauscher reaction, the Bray-Liebhafsky reaction and the iodine clock reaction. In that setting, their importance was recognized pretty early [13, 14, 29].

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More recently, in various models of biological pattern formation of Turing type, for the purpose of replicating experimental results in early 1990s, whether it is CIMA or Gray-Scott [20,22] chemical reaction of the form

$$A + 2B \rightarrow 3B$$
 and  $B \rightarrow C$ ,

with *C* an inert chemical species, plays a significant role. In particular, in Gray-Scott model with feeding, self-replicating traveling pulse (traveling wave) is the most exciting and not completely understood phenomenon [9–11, 18].

In this work, we study the traveling wave problem of autocatalytic chemical reaction  $A+nB \rightarrow (n+1)B$ , which, after simple non-dimensionalization results in the reaction-diffusion system,

$$(I) \begin{cases} \frac{\partial u}{\partial t} = \frac{\partial^2 u}{\partial x^2} - uv^n, \\ \frac{\partial v}{\partial t} = D\frac{\partial^2 v}{\partial x^2} + uv^n, \end{cases}$$

as well as that of chemical reaction  $A+nB \rightarrow (n+1)B$ , and  $B \rightarrow C$ , which has the governing equations

$$(II) \begin{cases} \frac{\partial u}{\partial t} = \frac{\partial^2 u}{\partial x^2} - uv^n, \\ \frac{\partial v}{\partial t} = D\frac{\partial^2 v}{\partial x^2} + uv^n - kv^m \end{cases}$$

Here, *D* a positive constant is the ratio of diffusion coefficients of chemical species *B* to that of *A*,  $n \ge 1$  is a positive constant not necessarily an integer, and  $kv^m$  describes the rate of  $B \rightarrow C$ , with *k* and  $m \ge 1$  both positive constants. We assume throughout that  $1 \le m \le n$ .

For a traveling wave solution to (I), u(x,t) = u(z), v(x,t) = v(z), where z = x - ct, the governing ODE system is:

$$\begin{cases} u'' + cu' - uv^n = 0, \\ Dv'' + cv' + uv^n = 0, \end{cases}$$
(1.1)

where c > 0 is a constant. Assuming

$$\lim_{z \to -\infty} (u, v) = (0, a), \qquad a > 0,$$

the addition of the two equations and integration on  $(-\infty, z]$  yield

$$u' + Dv' + c(u + v - a) = 0.$$
(1.2)